

Evolution of Structure and Optoelectronic Properties During Halide Perovskite Vapor Deposition

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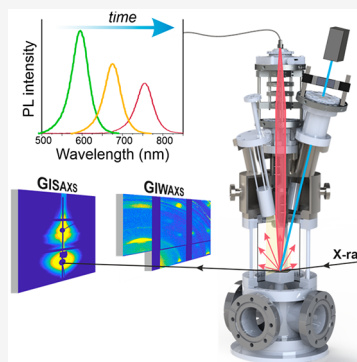


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Supporting Information

ABSTRACT: The efficiency of perovskite-based solar cells has increased dramatically over the past decade to as high as 25%, making them very attractive for commercial use. Vapor deposition is a promising technique that potentially enables fabrication of perovskite solar cells on large areas. However, to implement a large-scale deposition method, understanding and controlling the specific growth mechanisms are essential for the reproducible fabrication of high-quality layers. Here, we study the structural and optoelectronic kinetics of MAPbI₃, employing *in-situ* photoluminescence (PL) spectroscopy and grazing-incidence small/wide-angle X-ray scattering (GI-SAXS/WAXS) simultaneously during perovskite vapor deposition. Such a unique combination of techniques reveals MAPbI₃ formation from the early stages and uncovers the morphology, crystallographic structure, and defect density evolution. Furthermore, we show that the nonmonotonous character of PL intensity contrasts with the increasing volume of the perovskite phase during the growth, although bringing valuable information about the presence of defect states.



The increasing demand for electric energy requires more efficient and cost-effective technologies that are less technically challenging. Here, solar cells based on organic lead halide perovskites have shown great potential as an emerging photovoltaic technology.^{1–4} This relatively new material, possessing high absorption, tunable direct bandgap, and extensive compositional engineering possibilities, has made remarkable progress with ever-increasing power conversion efficiency (PCEs) exceeding 25%.^{5–8} As apparent by their notation, the organic lead halide perovskite films are formed by a chemical reaction between at least two precursors—lead halide, such as PbI₂ or PbCl₂, and organic halide, e.g., CH₃NH₃I, CH₃NH₃Cl, or CH₃IN₂ (MAI, MACl or FAI, respectively). Numerous fabrication methods have been developed to prepare perovskite layers, with the solution-processed spin coating being the most commonly used. Although the spin coating is advantageous for basic laboratory research due to its rapid manufacturing processes and low equipment costs, it is not suitable for large-scale industrial/commercial production.^{9–12} One of the reasons for this is a high material waste rate of over 90%.¹³ Furthermore, poor control of crystallinity and stoichiometry occurs in the case of the incomplete reaction of precursors during annealing, leading to poor batch-to-batch reproducibility.¹⁴

The vapor-based deposition technique, where the solid precursors are thermally evaporated onto the substrate in a high vacuum, provides several advantages over wet processes.¹⁵ This technique allows the fabrication of high-purity layers, where the film thickness can be precisely controlled on a large scale. Moreover, in view of the recent advances in high-

efficiency perovskite/silicon tandem solar cells,^{16,17} it provides a tool for precise codeposition of multiple layers without affecting the underlying substrate. Thus, vapor deposition is a perspective method, especially for multijunction or all-perovskite tandem solar cells.^{18,19}

Since vapor deposition, requiring dedicated/specialized high-vacuum deposition chambers, is not an extensively utilized technique for perovskite fabrication, the majority of published work is focused on the specific deposition parameters and their influence on the perovskite layer quality.^{11,12,20–22} The main cause is the nontrivial evaporation of the organic precursor, typically MAI, that has a low sticking coefficient influencing its adsorption onto the substrate.^{20,21} However, to the best of our knowledge, a comprehensive understanding of the perovskite evolution during vapor deposition in terms of morphology and electronic and structural properties is missing. To obtain an unbiased picture during perovskite formation, noninvasive *in-situ* experimental techniques need to be employed. Interrupting growth to perform various *ex situ* measurements under ambient conditions can result in misleading data, as perovskites are sensitive to humidity and light exposure.²³

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Here we present a simultaneous measurement of *in-situ* photoluminescence (PL) and grazing-incidence small/wide-angle X-ray scattering (GI-SAXS/WAXS) that reveals the electronic and structural changes during vapor-based growth of the archetypal MAPbI₃ perovskite. Such a unique combination of measuring techniques enables observation of perovskite phase formation from early stages and reveals the time evolution of defect density, morphology, and crystallographic structure. We show that perovskite formation is similar to wet deposition despite a different growth method. Specifically, the overall PL intensity is not directly proportional to the perovskite volume phase. The nonmonotonous character of PL was observed during growth, indicating the formation of defect states at the grain boundaries. We believe that the combined PL and GI-SAXS/WAXS techniques presented here are of fundamental importance for applying advanced passivation techniques that would lead to efficient perovskite layers beneficial for various optoelectronic applications.

We performed a vapor deposition of MAPbI₃ perovskite, where PbI₂ and MAI precursors were evaporated from the crucibles onto an unheated Si substrate (with native oxide) in a vacuum chamber with a base pressure of $\sim 10^{-6}$ mbar. Once the precursors' temperature was stabilized, the shutters on both crucibles were opened simultaneously, and the perovskite layer started to form. The *in-situ* PL and GI-SAXS/WAXS were measured simultaneously during the MAPbI₃ formation, as schematically outlined in Figure 1. A detailed description of the experimental conditions and setup can be found in Supporting Information and ref 24.

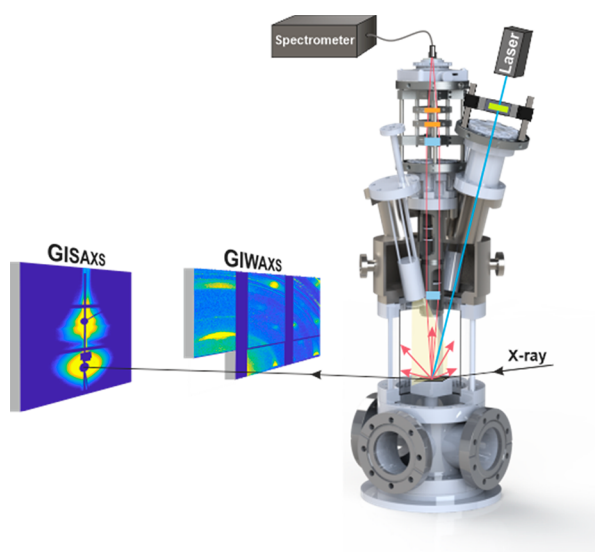


Figure 1. Schematic drawing of the deposition chamber and the scattering geometry.

To thoroughly understand the processes during perovskite growth, the surface morphology needs to be investigated with a temporal resolution faster than the progressing structural changes. Interrupting the perovskite deposition to acquire real-space images at different growth stages could be problematic, since it typically requires exposing the perovskite layers to ambient conditions. Perovskites are very sensitive to air or/and humidity,^{1,23,25,26} so we used *in-situ* GISAXS to monitor the surface morphology noninvasively during the layer formation.

Figure 2a shows the GISAXS diffraction pattern measured ~ 2 min after opening the shutters on PbI₂ and MAI. Here, diffuse out-of-plane scattering can be observed (indicated by magenta dashed lines in Figure 2a), containing information about the perovskite layer morphology. The diffuse scattering rods along $q_{\perp}$ indicate laterally ordered nanostructures at the beginning of perovskite growth. We confirmed the existence of such perovskite nanostructures at the beginning of perovskite deposition by taking postgrowth AFM images (see Figure 2b). The AFM scans were done on a perovskite layer where the growth was interrupted ~ 2 min after the deposition started. The AFM image shows isolated perovskite islands with an average particle size distribution $d_{\text{AFM}} = (90 \pm 10)$ nm and an average height of ~ 5.5 nm. The mean interisland spacing was determined by calculating the radial power spectral density function (PSDF) and fitting it subsequently by a Gauss function (see Figure 2c). The maximum of the fit is at ~ 0.023 nm⁻¹, corresponding to the mean distance between islands $D_{\text{AFM}} \approx 275$ nm. We note that more AFM scans, including for thicker MAPbI₃ layers, are shown in Figure S1.

Similarly, the mean distance between the islands can be deduced from the lateral patterns in the GISAXS scans. The reciprocal distance $2\Delta q_{\parallel}$ between the out-of-plane peaks (shown in Figure 2a) gives the average correlation distance between the perovskite islands $D \approx 2\pi/\Delta q_{\parallel}$.^{27,28} The time evolution of the out-of-plane scattering peaks, obtained by integrating the reciprocal area in the vicinity of $q_{\perp} \approx 0.09$ Å⁻¹, is shown in Figure 2d. At the beginning of the deposition, the peak position is at $q_{\parallel} \approx \pm 0.017$ nm⁻¹ and continuously shifts toward lower values as the perovskite growth proceeds. Such evolution of roughness spatial frequencies (represented in this case by the position of the out-of-plane peaks along $q_{\parallel}$) from larger values toward lower ones is typical for mounded surfaces during vapor deposition.²⁹ With increasing deposition time, the film roughness at the higher spatial frequencies is reduced faster than the one at the lower spatial frequencies, implying faster decay of the scattered intensity at higher $q_{\parallel}$. Similar evolution of the out-of-plane scattering peaks from larger $q_{\parallel}$ toward lower values could be observed during the island coalescence, caused by an increase in the mean distance between the nanostructures due to increasing island size.^{27,30,31} However, the island coalescence typically appears at much higher spatial frequencies ($q_{\parallel}$). In our case, the continuous decrease of $q_{\parallel}$ during the deposition is related solely to mounded growth as the island coalescence appears ~ 3 min after the deposition, as will be discussed in more detail later in the text.

The calculated distance between perovskite islands from the reciprocal distance $2\Delta q_{\parallel}$ between out-of-plane peaks at the beginning of growth is $D \approx 370$ nm. This value is close to the mean distance obtained by AFM ($D_{\text{AFM}} \approx 275$ nm). The difference between these two values can be explained by the fact that the overall scattering intensity is given by the product of the structure factor $S(q_{\parallel})$ and the form factor $F(q)$.^{31–33} $S(q_{\parallel})$ describes the spatial arrangement of the scattering particles (i.e., islands), and their correlation and $F(q)$ is given by the particle shape. In our previous calculation, we calculated D directly from the position of the out-of-plane diffraction, omitting the fact that its reciprocal position is also influenced by $F(q)$. The $F(q)$ -function of the nanostructures shifts the peak position to lower values, which might inaccurately indicate the larger mean distance between the perovskite islands. Nevertheless, the D value is important for

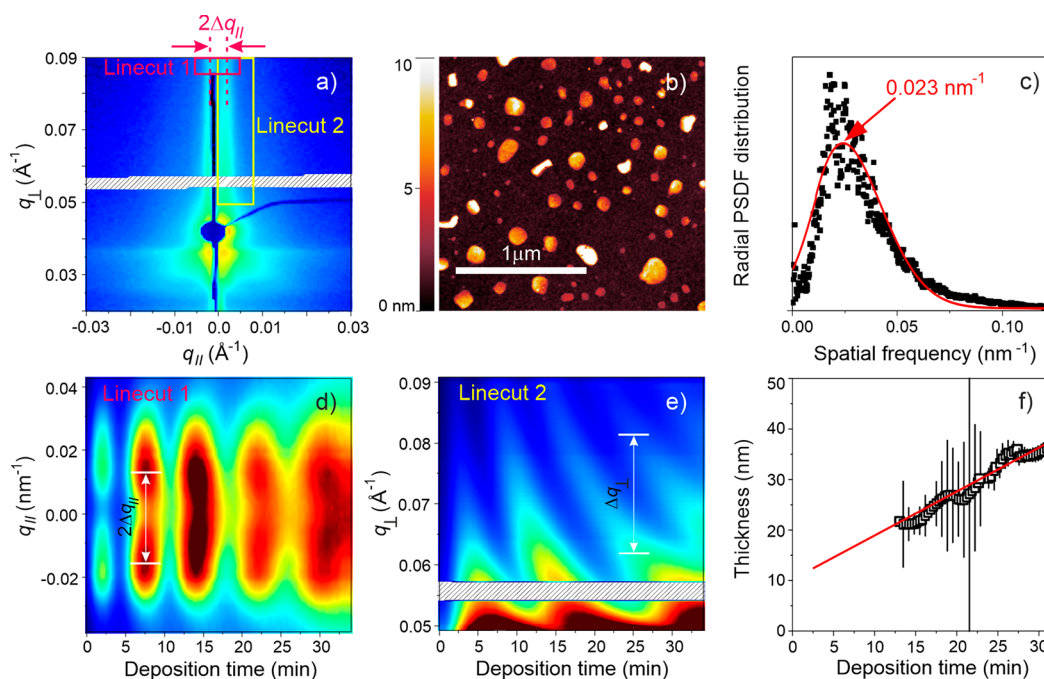


Figure 2. (a) GISAXS diffraction map containing information about the surface morphology. The reciprocal distance $2\Delta q_{\parallel}$ between the out-of-plane scattering patterns, indicated by magenta dashed lines, corresponds to the inverse average island spacing. (b) AFM image of 2D perovskite islands and (c) a corresponding power spectral density function (PSDF) distribution calculated from the AFM image. The red curve indicates the fit by a Gaussian function with a maximum at 0.023 nm^{-1} . (d) and (e) Diffuse scattering along $q_{\parallel}$ and $q_{\perp}$ directions during perovskite layer growth obtained from Linecut 1 and Linecut 2 indicated in (a), respectively. (f) Temporal perovskite layer thickness evolution, calculated from the reciprocal distance between two scattering intensity maxima indicated as $\Delta q_{\perp}$ in (e).

the final photovoltaic performance of the perovskite device since it defines the average grain size of the perovskite film. Several studies show that layers with large grains are more suitable for highly efficient photovoltaics.³⁴ It is worth noting that the grain size and overall morphology of the perovskite film are strongly affected by the substrate.^{35,36} For the vapor-deposited perovskites, a high density of nuclei was observed on various (inorganic) substrates, leading to a small grain size typical for coevaporated perovskite films. It was also shown that the use of organic substrates could enhance the morphology and overall optoelectronic properties of the perovskite film.³⁵

In addition, the GISAXS data provide information about the thickness of the deposited film. In conventional vacuum deposition, the layer thickness is typically deduced by monitoring the material deposited on the quartz crystal microbalance (QCM). However, this technique is not directly applicable to organic-based perovskites due to the volatile nature of the MAI precursor.^{10,12,20,37} However, the deposited layer thickness can be inferred from the intensity oscillations along the diffuse scattering rod in the $q_{\perp}$ direction. Figure 2e shows the oscillations along $q_{\perp}$ in time. These oscillations are caused by the interference effects and shorten their period ($\Delta q_{\perp}$) due to the increasing thickness of perovskite film (same behavior as in Figure 2d). Analyzing the reciprocal distance between the intensity maxima for each deposition time, we were able to calibrate the perovskite film thickness, see Figure 2f, and the deposition rate of $\sim 1.4 \text{ nm/min}$. The layer thickness cannot be determined in the early stages of perovskite growth, as it does not form a continuous layer, but separated islands as discussed earlier. Within the accessible range in the $q_{\perp}$ direction, which is restricted by the experimental conditions, it was impossible to unambiguously

determine at which time the coalescence of the islands occurs. The layer thickness could be obtained from $\sim 12 \text{ min}$ after the deposition started. However, the moment of the coalescence—when the grains begin to touch—could be deduced indirectly from the PL measurements, as explained later in the text.

To investigate the evolution of the MAPbI₃ crystallographic properties, the GIWAXS patterns were recovered. Research on the vapor-based deposition of perovskites is not predominantly focused on the crystallographic variations of the structure during growth, presumably due to the need for a custom-designed system. In most studies, the diffraction patterns are used to determine the crystal phases, i.e., tetragonal or cubic, while changing external parameters, such as the working pressure, precursor temperature, or substrate temperature.^{12,20,37–40} Here, we investigated the changes in the crystallographic structure during MAPbI₃ growth in a vacuum.

Figure 3 shows the GIWAXS reciprocal space map obtained at the end of the MAPbI₃ deposition (layer thickness $\sim 37 \text{ nm}$). The positions of the diffraction peaks are consistent with the tetragonal crystal structure typically observed for MAPbI₃ at room temperature.^{41–43} Our GIWAXS pattern indicates good film quality with no excessive amount of any of the precursors observed. The perovskite layer is composed of monocrystalline domains (grains) that mostly exhibit one preferential crystallographic orientation with the (001) lattice planes parallel to the substrate surface. This feature is in stark contrast to the grain orientation obtained for solution-based techniques, where the grains are typically randomly oriented.⁴⁴ Furthermore, we have chosen the 002 and 110 diffraction peaks and analyzed their evolution during perovskite deposition by fitting with Gaussians (see Figure S2). These diffraction peaks were selected because they are the most intense ones and thus can be analyzed from the early stages of MAPbI₃ growth.

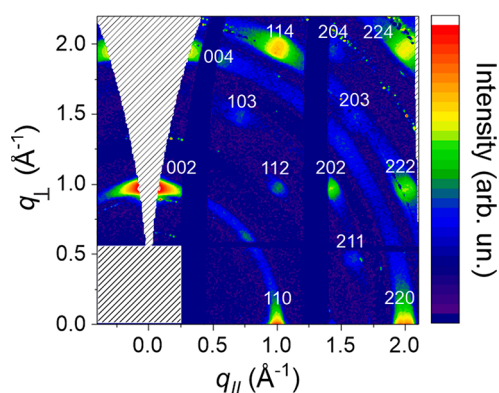


Figure 3. GIWAXS reciprocal diffraction map taken at the end of the MAPbI₃ growth (deposition time = 34 min, layer thickness ~37 nm). The diffraction peaks from the corresponding lattice planes are labeled. The detected scattering intensity is colored on a logarithmic scale to highlight less intense diffraction peaks. The color scale indicates the intensity increase from blue to red.

Additionally, 002 and 110 diffraction peaks provide information on all three lattice parameters a , b , and c ; i.e., the unit cell evolution can be studied in all three spatial directions.

Moreover, we aim to correlate the structural properties of the perovskite layer with the electronic ones. So before further investigating the structural properties obtained by GIWAXS, we now turn to results acquired from the PL measurements. Figure 4 shows the heat map of the PL intensity measured during MAPbI₃ deposition. The spectra were fitted by the PseudoVoigt function as it matched the spectral profiles well. The obtained spectral position of the PL peak, indicated as a dashed line in Figure 4a, continuously redshifts from higher energies ~1.95 eV at the beginning of the growth toward ~1.63 eV for thicker layers, which is a typical emission of evaporated MAPbI₃.^{21,44–46} The PL shift can be attributed to the strong quantum confinement effects at the beginning of

growth, where crystallites of a few nanometers are formed.^{47–49} According to the theoretical calculations by Anaya et al.⁴⁷ and Zhang et al.,⁴⁸ the PL peak position of 1.95 eV corresponds to a nanocrystallite size of ~6 nm. The emission continuously redshifts during the deposition due to the gradual enlargement of MAPbI₃ crystallites. The redshifting of PL during the layer formation was also observed for the solution-processed MAPbI₃,^{44–46,50} however, the total energy shifts did not exceed 0.15 eV. In contrast, here we observed an overall PL peak shift of ~0.32 eV, covering the emission from MAPbI₃ quantum dots to a closed layer. For comparison, Parrot et al.⁵¹ reported a redshift for coevaporated layers where the PL spectra were measured *ex situ* for various MAPbI₃ thicknesses. The PL peak was red-shifted ~0.1 eV from crystallite islands of a few-nanometer size to ~100 nm film thickness.

PL intensity analysis in Figure 4b shows a rapid emission increase at the beginning of perovskite growth up to its maximum value (~3 min after the deposition started), followed by a slower decrease. A similar PL intensity evolution was recently observed for solution-processed perovskites during annealing^{45,46,50} and assigned to the formation of surface defects and grain boundaries.⁴⁴ Despite the absence of the liquid phase in vapor-based deposition, the PL intensity character is the same for vapor and solution-based techniques. Based on the overall PL evolution, we can distinguish three different stages of MAPbI₃ crystallization.

Stage I: Formation of perovskite grains. In the first stage of perovskite deposition, small nanocrystallites are formed, continuously growing as more material is deposited (see Figure 4d). This stage is characterized by the PL intensity increase, reaching its peak at ~3 min of deposition. The redshifting of the PL peak position is fast due to the growth of perovskite crystallites in all three directions. The growth of the perovskite phase can also be confirmed by the increasing 002 diffraction intensity in Figure 4c. Interestingly, we observe a rapid change in the 002 peak position during this stage of

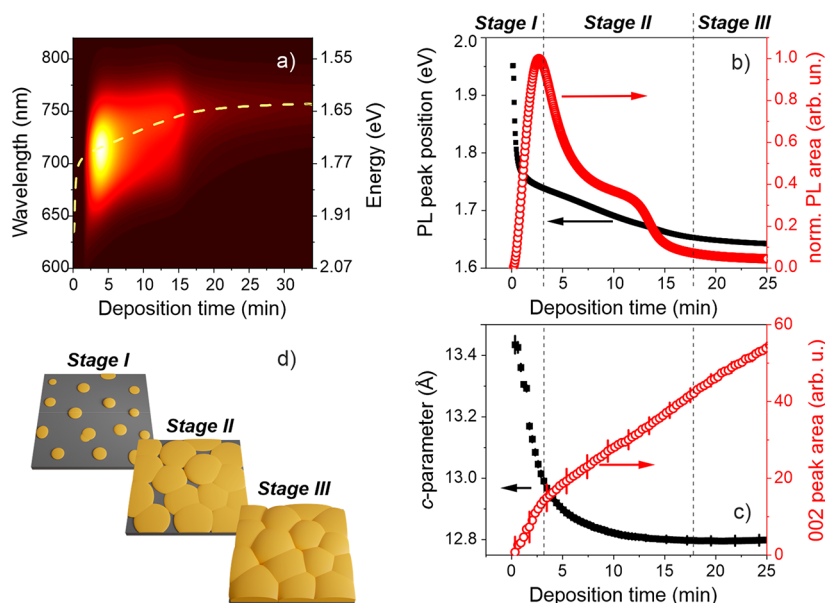


Figure 4. (a) Heat map of spectrally resolved photoluminescence (PL) obtained during MAPbI₃ deposition. The yellow dashed line indicates the PL peak position. (b and c) Comparison of selected parameters and their time evolution. (b) Time evolution of PL peak area and position and (c) GIWAXS 002 peak area and MAPbI₃ lattice parameter c . The gray dashed lines in (b) and (c) indicate the three different stages of MAPbI₃ growth that are described in the text and schematically depicted in (d).

perovskite growth (Figure S2), which represents a shrinkage of the unit cell c -parameter (see Figure 4c). A significant reduction of c can be explained by a strain induced by the rapid growth of perovskite grains. The overall decrease of the c -parameter is $\Delta c/c \approx 4.5\%$, with respect to its steady value at the end of the layer growth, where $c = 12.8 \pm 0.1$ Å. In our case, the growth-induced compressive strain acts mainly in the vertical direction and translates into shrinkage of the unit cell c -parameter, whereas a and b do not change substantially. We note that only a minor change in a and b of $\pm 0.3\%$, i.e., within the experimental error, was obtained from the reciprocal position of the 110 peak in Figure S2.

Previously, the origin of the strain was explained by the thermally induced cubic-to-tetragonal phase transitions in solution-based perovskites.^{52,53} For solution-based films, the high processing temperatures of ~ 100 °C are used, and the perovskites undergo such a phase transition (at ~ 57 °C) during cooling to room temperature. In our experiment, the substrate was kept at room temperature throughout the deposition, so the cubic-to-tetragonal transition as the cause of the strain is very unlikely. Additionally, no transition in the GIWAXS pattern was observed during perovskite growth.

Stage II: Formation of defects. This stage is characterized by a significant PL quenching. Such a PL intensity decrease was observed previously during perovskite formation,^{45,46,50,54} but its origin was not fully explained. Song et al.⁴⁶ speculated that the gradual drop of PL intensity is caused by photodarkening and thermally induced effects. However, the thermally induced PL quenching is unlikely, as the intensity of the laser diode in our experiment was set low to overcome the thermally driven bleaching effects, and although the photodarkening effects were previously observed in optical fibers,⁵⁵ no relevant evidence was provided to support the assumption of this hypothesis in perovskites.

We suggest that the PL quenching can be explained by the coalescence of individual perovskite crystallites. The coalescence was confirmed by AFM images of the MAPbI₃ layer obtained at the time when the PL intensity had just started to decrease (see Figure S1). This assumption is also accompanied by a slower redshift of the PL peak than in Stage I as the perovskite growth continues only in the direction perpendicular to the substrate plane. The grain boundaries between the crystallites act as defects and sources of nonradiative PL recombination.^{56–59} The formation of nonradiative defects is one of the major obstacles in achieving the highest possible power conversion efficiency, as they decrease the quasi-Fermi-level splitting that limits the open-circuit voltage of the solar cells.^{60–62} Thus, this stage (II) can be considered the most critical in perovskite film formation. Various passivation strategies, exploiting organic or inorganic molecules to precipitate at grain boundaries, can be applied to mitigate the nonradiative recombinations.^{63,64}

Recently, a correlation between growth-induced strain and nonradiative recombinations was observed by Jones et al.⁵² They observed that growth-induced lattice strain is directly linked with a greater defect concentration and hence with PL decay. A similar observation was also done by Jariwala et al., who focused on measuring local strain within a perovskite grain.⁶⁵ On the contrary, although our experiment allows measuring the strain averaged over a millimeter-sized area, such an unambiguous correlation between PL quenching and compressive strain cannot be made. On the one hand, the compressive strain, as evidenced by the c -parameter decrease,

is present in Stage II along with the PL intensity decrease. On the other hand, a significantly greater strain did not have a large effect on the PL amplitude during MAPbI₃ grain formation Stage I. Hence, our measurements suggest that the growth-induced compressive strain does not have a major influence on the overall PL signal.

Interestingly, the 002 diffraction intensity seems to slow down in Stage II (compared to Stage I), although the precursors' flux is approximately constant during deposition. However, this apparent intensity deceleration is caused by a change in the texture of the polycrystalline film. In Stage I, the perovskite islands are randomly oriented in the sample plane, having some orientation deviation in the out-of-plane direction. The out-of-plane misorientation is, in fact, a tilt of the crystallites' c -axes. Such tilt is evident in the broadening of the 002 diffraction peak in the in-plane direction (see Figure S3). After the crystallite coalescence in Stage II, the texture of the perovskite layer becomes more homogeneous, leading to a narrowing of the 002 peak. Consequently, the 002 peak coincides more with a *missing wedge* (see Figure 3), causing an illusive decrease in the peak intensity. In contrast, no such change in intensity slope was observed for the 110 diffraction peak because its position in reciprocal space is far from the *missing wedge*.

Stage III: Steady perovskite layer growth. The final stage of MAPbI₃ growth is characterized by a steady PL peak position (~ 1.63 eV) and a constant value of the c -parameter (see Figure 4b,c). Such stable growth was observed after ~ 17 min of the deposition. Although MAPbI₃ deposition continues with an almost steady flux of the precursors, the PL and GIWAXS intensities have different behaviors. The GIWAXS intensity keeps increasing monotonically, while the PL signal remains constant. Such distinct behavior is caused by different absorption coefficients (and thus penetration depths) for the 405 nm excitation wavelength and the X-ray beam. For the X-rays ($\lambda_{\text{X-ray}} = 1.05$ Å; $E_{\text{X-ray}} = 11.83$ keV), the penetration depth is ~ 50 nm. So, unless the perovskite layer thickness is not greater than 50 nm (for the X-ray incident angle of 0.2° , see Experimental methods in the Supporting Information), the GIWAXS intensity will increase as more material is deposited. However, for the excitation laser ($\lambda_{\text{ex}} = 405$ nm; $E_{\text{ex}} = 3.06$ eV), the penetration depth is only ~ 30 nm.^{66–68} We calculated that the MAPbI₃ thickness is (26 ± 4) nm after 17 min of deposition (see Figure 2f). Thus, from this moment on, the laser excites the same perovskite volume despite the increasing layer thickness, and the PL intensity remains constant.

In summary, we studied the formation of MAPbI₃ perovskite employing vapor deposition. The development of MAPbI₃ from a few nanometer-size crystallites up to a closed layer was characterized in terms of structural and optoelectronic properties by using *in-situ* GI-SAXS/WAXS and PL simultaneously. We observed that the growth of MAPbI₃ nanocrystals is manifested in rapid PL intensity increase and a substantial PL redshift. At the same time, the growth-induced compressive strain leads to a rapid decrease of the unit cell volume, predominantly given by the shrinkage of the c -parameter. Contrary to previous reports, we identified that the subsequent strain release does not influence the nonradiative recombination significantly. Furthermore, the formation of a continuous perovskite layer results in the development of defects on the grain boundaries, which quenches the PL signal. Interestingly, the PL evolution is remarkably similar to MAPbI₃ formation from solvent precursors observed in refs 44, 45, and 54. We

believe that further work is required, especially for vapor-deposited perovskites, to understand the complex connection between structure and emission efficiency. Such research is inevitable for optimizing the perovskite growth process and achieving highly efficient films on a large scale.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jpclett.2c03422>.

Experimental methods, AFM images, and complementary PL data and GI-SAXS/WAXS data (PDF)

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Notes

The authors declare no competing financial interest.

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